STUDIES ON A NEW ANTITUMOR ANTIBIOTIC, LARGOMYCIN. II ISOLATION, PURIFICATION AND PHYSICOCHEMICAL PROPERTIES

Totaro Yamaguchi, Tatsuo Kashida, Katsuyuki Nawa, Takehiko Yajma, Toshikazu Miyagishima, Yukio Ito and Tomoharu Okuda

Microbial Chemistry Research Laboratory, Tanabe Seiyaku Co., Ltd., Toda, Saitama, Japan

NAKAO ISHIDA and KATSUO KUMAGAI*

Department of Bacteriology, Tohoku University, School of Medicine, Sendai, Japan * Department of Microbiology, Tohoku University, School of Dentistry, Sendai, Japan

(Received for publication May 13, 1970)

Isolation and purification of largomycin from the culture filtrate of *Streptomyces pluricolorescens* MCRL-0367 is described. The crude complex of largomycin was obtained from the culture filtrate by means of precipitation with ammonium sulfate. Three active components were separated by the following procedures: gradient extraction with aqueous ammonium sulfate solution, isoelectric point precipitation, preparative electrophoresis on poly-acrylamide gel, AE-collulose column chromatography, gel-filtration on Sephadex G-100 and dialysis with a cellophane membrane. As a result, the most biologically active component, largomycin F-II, was obtained, the homogeneity of which was proved by ultracentrifugation and electrophoretic analyses. Largomycin F-II proved to be a chromoprotein of high molecular weight (25,000) containing a pH-indicator chromophore in the molecule. The most unique characteristic of largomycin F-II, which differentiate it from similar antibiotics is the lack of hexosamine and cysteine in its molecule.

In the course of screening for antitumor antibiotics, a new macromolecular antibiotic complex named largomycin was isolated from the culture filtrate of *Streptomyces puricolorescens* MCRL-0367. Taxonomic studies of the antibiotic-producing strain and antibiotic production have been reported in a previous paper¹). The present paper gives details of the isolation and purification, which made it possible to separate the largomycin complex into three active components named largomycin F-I, F-II, and F-III, and also gives the physichochemical properties of each component. Antitumor activity against transplantable animal tumors and biological characteristics of the antibiotics will be dealt with in the following paper².

Isolation and Purification

As will be mentioned below, largomycin and its active components are macromolecular proteinaceous substances with an unidentified chromophoric moiety, and

 $(\mathrm{NH}_4)_2\mathrm{SO}_4$

are relatively stable at neutral and weakly acidic pH. Considerring these properties, the isolation of largomycin from the culture broth was carried out as follows: The harvested broth was filtered with 2 % Celite 545, and ammonium sulfate added to the filtrate to give 0.7 saturation. The filtrate was kept overnight at 4°C to yield an active precipitate which was collected by filtration. The precipitate thus obtained was dissolved in distilled water and dialysed against water with a cellophane membrane. The inner solution was lyophilized to yield crude largomycin complex.

As shown in Fig. 1, when the crude largomycin complex was analyzed by gel-filtration on Sephadex G-100 equilibrated with M/15 phosphate buffer, three major peaks which were named F-I, F-II and F-III were observed. It was noted that the protein and sugar contents exactly run parallelled antimicrobial, anti-HeLa cell and antitumor activities. Of the three fractions, F-II showed the highest biological activity.

The largomycin complex was successfully separated into the three components, F-I, F-II and F-III, by applying the following procedures in succession.

- Gradient extraction with aqueous ammonium sulfate solution³⁾.
- 2) Isoelectric point precipitation
- Gel-filtration on Sephadex G-100





estimated: about 3,690 g)						
	extracted	with	0.7	satd.		

Extract Res	idue			
dialyzed	extracted with 0.5 satd. (NH ₄) ₂ SO ₄ .			
Inner solution	pptd. with (NH ₄) ₂ SO ₄ , filtered			
lyophilized gel-filtration	Hyflo-super-cell			
Largomycin H	-II and F-III mixture			
Largomycin F-I powder 1.0 g Res	washed with 0.7 satd. $(NH_4)_2SO_4$ due			
	extracted with 0.53 satd. (NH ₄) ₂ SO ₄			
Ext	ract			
	pptd. with (NH ₄) ₂ SO ₄ filtered with Hyflo-super-cell			
Yellowish powder				
	extracted with 0.5 satd. $(NH_4)_2SO_4$ - $M/10$ acetate buffer mixture (1 : 1, pH 4.0)			
Extract Resi	due			
dialyzed lyophilized gel-filtration	extracted with 0.5 satd. $(NH_4)_2SO_4$ - $M/10$ acetate buffer mixture (1 : 1, pH 6.0)			
Largomycin F-III Ext	ract			
powder 1.0 g	dialyzed, lyophilized, gel-filtration			
Largomycin F-II				



As shown in Chart 1, largomycin crude complex obtained from the broth filtrate was first extracted with 0.7 saturated ammonium sulfate solution* to separate the F-I component from the insoluble F-II and F-III components. Largomycin F-I was then dialyzed, lyophilized, and purified through gel-filtration. Thus, largomycin F-I powder was obtained. The F-II and F-III components, which remained undissolved in 0.5 saturated ammonium sulfate solution were extracted with 0.5~0.53 saturated ammonium sulfate solution to free them from other impurities. The yellowish powder recovered from the extract was then treated with pH 4.0 acetate buffer solution and an equal volume of 0.5 saturated ammonium sulfate. Largomycin F-III was soluble and was thus removed from the insoluble F-II component. Largomycin F-III in the extract was recovered as a brown powder by dialysis, lyophilization, and gel-filtration. Largomycin F-II was extracted with a 1:1 mixture of pH 6.0 acetate buffer solution and 0.5 saturated ammonium sulfate, and recovered as a light brown powder.

Sephadex G-100 gel-filtration analyses of the samples thus obtained revealed that the each component was still contaminated with small amounts of the other fractions. However, repetition of Sephadex G-100 gel-filtration of each component yielded components F-I, F-II and F-III in an almost pure state.

Chromatography utilizing an ion-exchanger was the most suitable procedure for further purification of largomycin F-II. The light brownish powder containing impure largomycin F-II (1 g) was dissolved in 0.01 M phosphate buffer (pH 7.0) and dialysed in a cellophane tube against the same buffer overnight in a cold room. The inner solution was applied to an AE-cellulose column (2.4×47 cm) which had been equilibrated with 0.01 M phosphate buffer (pH 7.0). The column was first washed completely with the same buffer and then developed with a linear gradient of phosphate buffer (pH 7.0) of concentrations ranging from 0.01 to 0.2 M. Eluates were scanned at 280 m μ and the sugar content in each eluate was determined by the phenol-sulfuric acid method⁵⁰ at 470 m μ . As shown in Fig. 2, the fractions which were eluted with buffer of around 0.07 M showed a maximum absorbancy at 280 m μ and had an O.D.₂₈₀/O.D.₄₇₀ ratio of 7~10. These fractions (A₁-III) were combined and dialysed in a cellophane tube against water overnight in a cold room. The



* Ammonium sulfate solutions used in the present experiment were prepared by HOFFMEISTER'S method.

_.

Fig. 4. Gel-filtration of largomycin F-II on Sephadex G-100 column.



dialyzed solution was refractionated by the same AE-cellulose grachromatography dient $(1.5 \times 57 \text{ cm})$ (Fig. 3). The fractions which showed maximum absorption at 280 m μ (A₂-II) were combined and lyophilized without dialysis. The lyophilized (A₂-II) fraction was then redissolved in 5 ml of water and chromatographed with water on Sephadex G-100 in a 2.4×47 cm column (Fig. 4). Pure largomycin F-II was recovered by lyophilizng fractions from the second peak $(A_2-II-S-II)$ with the maximum absorption at 280 m μ . A₂-



Fig. 5. Isoelectric focussing pattern of largomycin F-II.







II-S-I fraction was less biologically active and contained minor components which could be detected by electrophoretic analyses in polyacrylamide gel⁶). The last component, A_2 -II-S-III, contained strongly bound phosphate which could not be removed by repeated gel filtration chromatography on Sephadex G-100 and G-75. The last component, however, showed no biological activity.

Further purification of largomycin F-II fraction (A_2 -II-S-II) was accomplished by preparative electrophoresis using polyacrylamide gel*. The acrylamide gel column was extruded from the apparatus, and the zone containing the active principle was cut into small pieces and extracted with distilled water. The active extract was

^{*} This apparatus was devised by one of the authors (T. KASHIDA) in our laboratory and will be described elsewhere.

dialysed against water and lyophilized to yield a powder, which was again subjected to gel filtration, dialysis and lyophilization. The final product was a pale yellow amorphous powder, the homogeneity of which was proven by electrofocussing analysis⁴⁾ and also by ultracentrifugation as shown in Figs. 5 and 6 respectively.

Physicochemical Properties

Physicochemical properties of largomycin F-I, F-II and F-III are summarized in Table 1. Whereas largomycin F-II was homogenous, largomycin F-I and F-III were still contaminated with minor impurities.

Purified largomycin F-II is an acidic protein with an isoelectric point at pH 4.2^{4} (Fig. 5). The molecular weight of largomycin F-II was estimated to be about 25,000 by the ARCHIBALD method,^{7,8)} and from amino acid analysis. The value was also supported by the sedimentation constant of about 2.2 S and by gel-filtration utilizing ANDREW's procedure⁹⁾.

As shown in Fig. 7, the infrared spectra of largomycin components in KBr tablets were quite similar to each other and were typical of a polyamino-polyoxy compound. As shown in Fig. 8, the ultraviolet absorption spectrum of largomycin F-II showed a maximum at 278 m μ ($E_{lem}^{1\%}$ 19.4) in 0.1 N HCl and 540 m μ ($E_{lem}^{1\%}$ 4.0) in 0.1 N NaOH solution. The latter suggested the presence of a chromophoric moiety in largomycin F-II molecule.

The pH-indicator constituent in largomycin F-II which could not be removed by Sephadex chromatography or dialysis in cellophane tube against distilled water was

Properties	Largomycin F-I	Largomycin F-II	Largomycin F-III	
Nature	Nature Acidic, pale yellow amorphous powder		Brown amorphous powder	
Molecular weight	50, 000*	25, 000	20,000~10,000*	
Melting point (decomp.)		no definite		
Isoelectric point		pH 4.2		
Elementary analysis, found (%)	$\begin{array}{rrrr} {\rm C}:&47.93\\ {\rm H}:&7.46\\ {\rm N}:&6.50\\ {\rm S}:&0.82 \end{array}$	$\begin{array}{c} 47.83 \\ 8.30 \\ 13.63 \\ 0.25 \end{array}$	$\begin{array}{c} 46.79 \\ 6.57 \\ 8.64 \\ 1.08 \end{array}$	
UV absorption max. in 0.1 א HCl max. in 0.1 א NaOH	$275~{ m m}\mu$	$\begin{array}{c} 278 \ \mathrm{m}\mu \\ 540 \ \mathrm{m}\mu \end{array}$	$\begin{array}{c} 275 \mathrm{m}\mu \\ 425 \mathrm{m}\mu \end{array}$	
Sugar content (Phenol-H ₂ SO ₄)	56±1 %	3±1 %	22±1 %	
Sugar identified	Glucose	Glucose	Glucose	
Amino acid analysis (F-II)	Lys, Arg, His, Asp. Glu, Thr, Ser, Pro, Gly, Ala, Val, Leu, Ileu, Try, Tyr, Phe, Met			
Color reaction Positive	Ninhydrin, Biuret, Xanthoprotein, SAKAGUCHI, MOLISCH, Anthrone, Phenol-H ₂ SO ₄ , FOLIN-CIOCALTEU			
Negative	Elson-Morgan			
Precipitation	PPtn. with 50 % (NH ₄) ₂ SO ₄ , 50 % MgSO ₄ , 50 % ZnCl ₂ , 20 % T.C.A., 5 % rivanol			

Table 1. Physicochemical properties of largomycin

* These values were obtained by Sephadex gel filtration method.



extracted with glacial acetic acid from F-II as in the case of plurallin¹⁰.

The glacial acetic acid extract was dried in vacuo to yield pigment mixtures which were dissolved in chloroform and the chloroform solution was extracted with acidic water (pH 5.0, HCl). The aqueous layer was lyophilized to yield the crude hydrochloride of the pigment which was again dissolved into a small amount of chloroform. The chloroform solution was applied to an alumina column, developed with chloroform – methanol (95:5). The eluates were tested for activity

Fig. 8. Ultraviolet absorption spectra of largomycin F-II.



against HeLa cells and those in a single peak were combined and evaporated *in vacuo* to yield a yellowish brown hydrochloride. To obtain the free base of the pigment, the aqueous solution of the hydrochloride was extracted with chloroform at pH 7.5. The chloroform layer was concentrated *in vacuo* and precipitated with petroleum ether to yield a yellowish powder of the free base. This did not show any clear melting point up to 270°C. The ultraviolet absorption spectrum showed absorption maxima at 244 m μ (E^{1%}_{1cm} 556), 275 m μ (E^{1%}_{1cm} 380) and 433 m μ (E^{1%}_{1cm} 87) in methanol or methanol – 1 N HCl (9:1) solution and 255 m μ (E^{1%}_{1cm} 151) and 550 m μ (E^{1%}_{1cm} 56) in methanol – 1 N NaOH (9:1) solution (Fig. 9). In the infrared absorption spectrum, the major bands were at 3390, 2920, 1660, 1470, 1325, 1270, 1090 and 850 cm⁻¹ (Fig. 10). Elementary analysis gave following values: C 62.83, H 6.58, N 3.23 %. Thin-layer chromatography on silicagel with the solvent system EtOH – NH₄OH – H₂O (4:1:1) and on aluminum oxide with the solvent system *n*-BuOH – MeOH – H₂O (4:1:2), gave Rf values of 0.95 and 0.74 respectively. The pigment showed anti-HeLa cell activity at a concentration of 0.2 mcg/ml and also antitumor activity against EhrLich



* The content of tryptophan was determined by a colorimetric method¹³⁾ and an ultraviolet absorption method¹⁴⁾.





ascites carcinoma at 1.6 mg/kg. In infrared and ultraviolet spectra, the chromophore pigment of largomycin F-II was similar to that of iyomycin B_1 ¹⁵⁾. Although some discrepancy was observed on elementary analysis, the difference between largomycin pigment and iyomycin B_1 is not certain in this moment.

Table 2 illustrates the amino acid composition of largomycin F-II determined after hydrolysis with constant boiling hydrochloric acid at 110°C for 24 hours in an evacuated tube. The content of tryptophan was determined by a colorimetric method¹³⁾ and by ultraviolet absorption¹⁴⁾. It was characteristic that cysteine was not detected in largomycin F-II hydrolyzate. Amino acid analyses were also attempted on the hydrolyzates of largomycin F-I and F-III fractions. In these cases, cysteine was not detected as well.

Hydrolyses to examine for neutral sugars and amino sugars were carried out with 0.24 N hydrochloric acid at 100°C for 18 hours and 4 N hydrochloric acid at 100°C for

4 hours respectively. The sugar components in the acid hydrolysate of largomycin F-II were analysed by paper chromatography* and by gas chromatography of pertrimethylsilyl derivatives.** Glucose and traces of galactose, mannose and xylose were found. However, no amino sugar was detected.

Discussion

Three antitumor components, largomycin F-I, F-II and F-III, were obtained from the culture filtrate of *S. pluricolorescens* MCRL-0367. The most biologically active component F-II was obtained in a pure form. The physicochemical properties of largomycin F-II suggest that it is a member of the chromoprotein group of antitumor antibiotics. From published data, plurallin¹⁰, iyomycin¹¹, and prunacetin¹² seem to be related to largomycin F-II that they contain a pH-indicator chromophore. However, prunacetin is apparently different from largomycin F-II in the physicochemical characteristics of its protein moiety and chromophore.

Since plurallin was isolated from the same species of Streptomyces as the largomycinproducing strain, the antibiotics were compared most carefully. They could be differentiated from each other in the following points. Unlike largomycin F-II, plurallin gave a positive ELSON-MORGAN test. After acid hydrolysis plurallin gave cysteine which was not obtained from largomycin F-II. Furthermore, the pH indicator pigment extracted from largomycin by the same method as used with plurallin had a different infrared spectrum from the indicator prosthetic group of plurallin.

Since the iyomycin A complex reported in the literature seems to be a mixture and no details are reported on the prosthetic group of this antibiotic, a precise comparison of these antibiotics is difficult at present. However, iyomycin A could be differentiated from largomycin F-II by the fact that largomycin F-II does not contain cysteine.

References

- YAMAGUCHI, T.; T. FURUMAI, M. SATO, T. OKUDA & N. ISHIDA: Studies on a new antitumor antibiotic, largomycin. I. Taxonomy of the largomycin-producing strain and production. J. Antibiotics 23: 369~372, 1970
- YAMAGUCHI, T.; M. SETO, Y. OURA, Y. ARAI, K. ENOMOTO, N. ISHIDA & K. KUMAGAI: Studies on a new antitumor antibiotic, largomycin. III. Biological properties and antitumor activity of largomycin F-II. J. Antibiotics 23: 382~387, 1970
- SOEJIMA, M.: Gradient extraction—its principle and some experimental examples. Protein, Nucleic Acid Enzyme. 9: 1235~1243, 1964
- 4) VESTERBERG, O. & H. SVENSSON: Isoelectric fractionation, analysis, and characterization of ampholytes in natural pH gradients. IV. Further studies on the resolving power in connection with separation of myoglobins. Acta Chemica Scandinavica 20: 820~834, 1966
- 5) DUBOIS, M.; K. A. GILLES, J. K. HAMILTON, P. A. REBERS & F. SMITH: Colorimetric methods for determination of sugars and related substance. Anal. Chem. 28: 350~356, 1956
- ORNSTEIN, L.: Disc electrophoresis-I, Background and theory. Ann. New York Acad. Sci. 121:321~349, 1964
- Archibald, W. J.: A demonstration of some new methods of determining molecular weight from the data of the ultracentrifuge. J. Phys. Chem. 51: 1204~1214, 1947
- GINSBERG, A.; P. APPEL & H. K. SCHACHMAN: Molecular-weight determination during the approach to sedimentation equilibrium. Arch. Bioch. Biophy. 65: 545~566, 1956.

*	Solvent systems:	BuOH - AcOH - H_2O (4 : 1 : 4), AcOEt : <i>n</i> -PrOH : H_2O (57 : 32 : 13), and AcOEt
		H_2O : pyridine (4:4:2).
	Spray reagents:	Benzidine, ninhydrin, TTC and ELSON-MORGAN reagents.
**	Apparatus:	Shimazu Model GC-1C
	Column:	1.5 % SE-30 on Chromosorb W, 1.5 m.
	Condition:	Carrier gas N ₂ , column temp. 250°C

- ANDREWS, P.: Estimation of the molecular weight of protein by Sephadex gel-filtration. Biochem. J. 91: 222~233, 1964
- 10) OGAWARA, H.; K. MAEDA, K. NITTA, Y. OKAMI, T. TAKEUCHI & H. UMEZAWA: An antibiotic, plurallin, consisting of pluramycin-like prosthetic group and a glycoprotein. J. Antibiotics, Ser. A 19:1~9, 1966
- NOMURA, S.; H. YAMAMOTO, A. MATSUMAE & T. HATA: Iyomycin A, new antitumor antibiotic from *Streptomyces*. I. Isolation and properties of iyomycin complex. J. Antibiotics, Ser. A 17:104~111, 1964
- 12) ARAI, T.; S. KUSHIKATA, K. TAKAMIYA, F. YANAGISAWA & T. KOYAMA: Prunacetin A, an antitumor antibiotic from Str. griseus var. purpureus (Syn. Str. californicus). J. Antibiotics, Ser. A 20: 334~343, 1967
- OPIENSKA-BLAUTH, J.; M. CHAREZINSKI & H. BERBEC: A new, rapid method of determining tryptophan. Anal. Biochem. 6:69~76, 1963
- 14) BEAVEN, G. H. & E. R. HOLIDAY: Ultraviolet absorption spectra of protein and amino acid. Adv. Protein Chem. VII: 319~386, 1952
- SANO, Y.; N. KANDA & T. HATA: Iyomycin, a new antitumor antibiotic from Streptomyces.
 III. Isolation and properties of iyomycin B. J. Antibiotics, Ser. A 17: 117~123, 1964